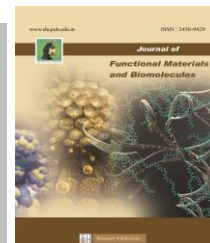




SACRED HEART RESEARCH PUBLICATIONS

Journal of Functional Materials and Biomolecules

Journal homepage: www.shcpub.edu.in



ISSN: 2456-9429

Comprehensive Review on Nanomaterial-Mediated Water Treatment strategies

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Received on 18 May 2021, accepted on 07 June 2021,

Published online on 21 June 2021

Abstract

Water pollution, with major economic and social costs, is a common problem affecting the different living organisms and environment. The increasing lack of available water supplies needs efficient wastewater treatment. Water treatment has become a growing problem nowadays. Its care has the chance to be in this progressive world. In wastewater treatment, nanomaterials have a staggering ability to be used. Any of its interesting features with more surface area can be successfully used to extract hazardous metal particles, microorganism-causing afflictions, daily solutes, and water inorganic. In this sense, the need for an hour is to develop cheaper, cleaner and more effective wastewater treatment technologies. The application of nanomaterials towards the water has been a promising strategy stated by several studies to be effective. This analysis offers nanobiotechnology advances that have an increased activity in bioremediation of wastewater. This investigation expressed nanomaterial mediated water treatment techniques, including nano filtration, membrane filtration, photocatalysis and adsorption with modern parameters offers to enhanced efficiencies and properties can be improved as needed.

Keywords: Nanoparticles, Water pollution, Wastewater, Nano Adsorption, Nanomaterials.

1 Introduction

The fundamental technique that nanotechnologies would benefit from is that the basic problems are instructive by expelling pollutants in water, including creatures such as pathogenic, pollution, insecure synthetic substances such as arsenic, mercury, pesticides, bug sprays and salt pose, etc., by and wide, rather than dumping, as ecological properties, then onto the next ¹. The scenario was compounded by the mismanagement of wastewater and the lack of public policy. It is, therefore, a dynamic and multi-dimensional problem. Researchers have built new technologies in the face of this situation to ease the burden on this scarce natural resource. Nanomaterials can offer a wide variety of uses due to their specific and novel surface area, such as reactant films, nano sponges, bio nanoparticles and metal nanoparticles, such as pinch, silver and TiO₂. As some master evaluations show, by 2050, the total population will grow to nine billion^{2,3}. Totally 77 percent of the national population is concentrated. The obtained

results were interpreted with Water Consultative Council reports which is considered as semiarid or 2/3 of territory monitoring.

The supply cost for drinking water for different localities was increasing day by day and utility of water was increasing⁴. The researchers trying to find an effective nano biomaterial for the elimination of toxins makes great attention among the different categories of peoples by different modified technologies. Among all different strategy'snanotechnology-based bioremediation of water offers increased impact to resolve all types of problems in the water treatment process^{5,6}.

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Nanotechnology covers multiple areas of research as diverse as molecular biology, molecular nanotechnology, semiconductor physics, organic chemistry, energy storage, surface science, molecular engineering and micro-manufacturing. In a variety of companies like healthcare, medicine, nano-EHS, climate, electricity and other technology in resolving processes. A main nanotechnology application is found in the segment of electronics and semiconductor devices, which is projected to rise by 2025 at a significant CAGR of 15.01 percent ⁷.

Business Growth Drivers for Nanotechnology:

The Application of Medical and Healthcare Nanotechnology

Wastewater treatment and other environmental aspects of the application

For Improved Clean Energy Nanotechnology

Business Dynamics for Nanotechnology:

Nanotechnology for Lighter but High-strength Materials-in the emerging technologies have been shown great attention towards carbon nanotube and graphene mediated devices offers great application. The recent trend is increased to express strong affinity and growth potential which can be seen in the lucrative market for graphene

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mediated nanoplatelets^{8,9}.

The improved insulating materials frequently used in the building industry is made up of a nanomaterial called aerogel that has silica nanoparticles. In addition, the cement production process contributes to the increased amount of CO₂ being evolved and by using nanotechnology it will ultimately reduce the emission of gases from the cement industry. In addition, it also contributes and makes a foot stem in glass industry, leather industry etc¹⁰.

BUSINESS PROBLEMS IN NANOTECHNOLOGY:

In the nanotechnology industry, the most difficult factor is production scalability. While nanomaterials provide excellent experimental performance in the research laboratories in model level, the scalability factor dwarfs the size of the market for nanotechnology¹¹. Therefore, some of the majorities of beneficial applications are in the research and development process. The major advances in the functioning of nanotechnology with commercials in the automobile, sporting goods and aerospace industries are anticipated in the future. The technology will also assist in the successful treatment of cancer, which will help manufacturers in the nanotechnology industry^{12,13}.

MARKET RESEARCH FOR NANOTECHNOLOGY SCOPE:

The study's baseline year is 2020, with projections for up to 2025. The report provides a detailed overview of the business environment, taking account of the developing industries' market shares. It also offers data on shipments by device. These exhibit the requisite business intelligence to the main market participants and help them understand the nanotechnology market's future¹⁴⁻¹⁶. The evaluation includes a forecast, a description of the competitive structure, competitors' market shares and market dynamics, market demands, market factors, market challenges and product analysis. In order to understand their effect over the forecast period, the market drivers and constraints were evaluated. The key growth opportunities are further described in these reports for chief challenges and potential intimidation¹⁷. The marketing strategies research study on nanotechnology also evaluates the use of nano devices, nano composites, nano equipment and different nano materials in different industries by product category¹⁸.

IN WASTEWATER TREATMENT NANOTECHNOLOGY

Nanotechnology has become one of the 21st century's most studied developments. Specific nanomaterial properties include:

- Large Ratio of Surface to Volume
- Small dimensions
- Well-organized framework
- Filtration Competence

Such characteristics help to eliminate heavy metals from contaminated wastewater. Wastewater treatment, depending on the form of nanomaterial, is divided into three major groups:

- Nanocatalysts Using
- Nano-adsorbent compounds

Nano-membranes

Nanocatalysts Using

Maximum focus has been given to photocatalytic technology for the control of water contamination. Photocatalytic operations involving the interaction of light energy with metallic nanoparticles are included in this procedure¹⁹⁻²². By reacting with hydroxyl radicals, the photocatalytic operation kills microorganisms (bacteria) and organic substances. Usually, inorganic materials such as metal oxides and semiconductors are the materials used in nanocatalysts²³.

NANO-ADSORBENT COMPOUNDS

Nanomaterials have two important characteristics that make them amazing adsorbents. They are a large surface of nanomaterial and are capable of reacting and binding to various bordering particles and molecules easily artificially. These characteristics make nanoparticles not only persuasive adsorbent for various contaminants in wastewater, but also contemplate the soundness of the entire deal, as this also results in adsorbent degradation and gains in adsorption profitability²⁴.

Organic or inorganic nanomaterials with a high affinity for the adsorption of substances are used in this treatment. Many pollutants are extremely capable of destroying these adsorbents. It has a large surface area, excellent catalytic potential, and high reactivity. The ideal adsorbent is thin²⁵. Nano-adsorbents, such as metallic nanoparticles, magnetic nanoparticles, nanostructured mixed oxides, and metallic oxide nanoparticles, are categorised on the basis of their adsorption method²⁶.

Nano-membranes

One of the main strategies to eliminate toxins from water is membrane filtration. Membranes offer a physical barrier to pollutants based on their size of the pore and size of the molecule. It is normal practice these days to use the reverse osmosis (RO) process to filter the non-potable water into portable water. Nanomembranes can isolate contaminants from wastewater during this treatment. These are used widely to eliminate heavy metals, colorants, and other pollutants²⁷. Nano membranes with various characteristics such as antimicrobial, selectivity, anti-fouling, enhanced permeability and photocatalytic can be developed depending on the application. The widely used nanomembranes are nanotubes, nanoribbons, and nanofibers. Silver nanoparticles are antimicrobial agents incorporated with high loads of bacterial pollutants during the treatment of wastewater. Silver nanoparticles and nanoparticles of graphene oxide play a dual role, i.e. prevention of biofouling (disabled bacterial cells), and they can minimise microbial attachment by forming a solid water layer due to their hydrophilic existence²⁸.

For industrial water treatment, widely used nanomaterials are discussed below:

Carbon (CNT) nanotubes:

Cylindrical structures consisting of graphene sheets with less than 1 nm of diameters are carbon nanotubes

(CNTs). The structure may be solitary or multilayer walled. In nature, they are having improved stability and duration. Reports demonstrate that the adsorption potential of carbon nanotubes to aromatic compounds is greater than that of metal cations. In emerging technologies CNT dependent fields are the removal of contaminants using nanocomposite materials shows great attention. CNTs create membranes that conduct electrically. The other type of nanocomposite material is magnetic carbon nanotubes (MCNTs). The pore size of the layer depends on the kind of polymers used to cross-link the CNT. MCNTs are well distributed in aqueous solution, making it simple to isolate and redevelop from the medium after treatment of contaminated water by using a magnet. The size of the membrane pore ranges from 100 nm to sub-nanometers. These nanomembranes are commonly used in a range of biological treatments for industrial wastewater and industrial brine desalination²⁹⁻³².

Nanoparticles of graphene oxide (GO):

The GO nanoparticles operate on the same principle as the tree leaf transpiration process, i.e., pulling water through the GO structure while catching solar heat successfully to stimulate vapour formation. In solar-powered brine desalination, this technique is used³³.

Nanoparticles with fluorinated silica:

A low-energy surface that has high resistance to oil is produced by fluorinated silica nanoparticles. This is particularly useful in the treatment of wastewater, oil refining, and food processing industries that produce oil³⁴.

Nanomaterials Layered:

In industrial water treatment, layered nanomaterials comprising graphene and graphene oxide, MoS₂, and MXenes (ultrathin transition metal carbonitride carbides and nitrides) are commonly used³⁵.

Existing and future uses for treating sewage and wastewater

Varieties of nanomaterials are commonly known as materials lesser than 100 nm. Products on this scale also have distinct dimensional characteristics which have been investigated in the treatment of wastewater³⁶. The slickly scalable nanomaterials properties of increased surface areas are used by many devices, including fast disintegration, strong sorption and elevated reactivities. Localized plasmon resonance, Superparamagnetic and quantum confinement have been further advantages of their discontinuous properties. It is possible to classify four major classes to be the majority of commonly used nanomaterials³⁷:

- a. Carbon-mediated nano-adsorbents in the treatment of waste water.
- b. In wastewater treatment, metal dependent nano-adsorbents.
- c. Polymer-dependent nano-adsorbents for the disposal of waste water.

d. Zeolite: Zeolite and its composites in the handling of waste water.

The following section briefly addresses each of these classes of nano-adsorbents.

Biologically organized membranes and nanocomposite thin films

The specificity and permeability of different biologically existing membranes possess determined protein channels which are known as aquaporins which are help water flow controlling factors of selective cell membranes and play critical role in the elimination of contaminants. Its improved selectivity and water permeability offer polymer membranes a quite attractive approach to enhancing membrane performance³⁸. *Escherichia coli* aquaporin Z when polymer vesicles are applied over the initial vesicles to the amphiphilic triblock water permeability with less glucose, glycerol, salts and urea rejection of magnitude. The coated significant lipid bilayers entrapped aquaporins on industrialized Nano filtration membranes is one of the potential designs³⁹.

The nano sized channels atomic sweetening architecture and a individual-files ordering of water through the nanotubes have been offers, aligned carbon nanotubes shows much faster movement of water than required by the Hagen Poiseuille Equations. In a membrane containing just 0.03 percent surface area of associated carbon nanotubes, it was systematically projected to have fluxes above current commercial RO marine based membranes⁴⁰. However, due to the absence of carbon nanotubes with a uniform sub-nanometer in diameter, it is very difficult to align the carbon nanotubes to reject salt or smaller molecules. Functional groups are gating on the opening of nanotube was suggested to improve the selectivity and sensitivity of aligning membranes of carbon nanotubes. By catching large functional groups at the opening of the channel, salt can physically be extracted. Steric exclusion, however, greatly reduces permeability. Therefore, at the present level, compatible membranes of carbon nanotubes cannot be desalinated. The diameter of the carbon nanotubes must be constant⁴¹.

Mechanisms of diverse nanoparticles mediated antimicrobial agents and possible applications in water treatment

Because of its special properties, including increased pollutant adsorption capacity, eminent photocatalytic and antimicrobial properties, researchers have paid effective attention to nanoparticles and enormous nanocomposites. By offering a broad range of antimicrobial pathways, nanoparticles disable microbial pathogens. Inorganic and organic contaminants are also isolated from wastewater by nanoparticles and reveal their potential use in wastewater management⁴²⁻⁴⁵. In potable, surface, drinking and other types of water supplies for inactive bacteria, TiO₂ is an extremely prevalent form of nanoparticle.

The antibacterial role of titanium dioxide is due to its production of reactive oxygen species. The ROS formed

can destroy the sensitive cytoplasmic membranes, damage proteins and DNAs, hazardous ions, disrupt electron flow, and interfere with the function of the respiratory system. Strong UV-A absorbance by solar radiation of titanium dioxide activation, dramatically increasing solar disinfection. Titanium dioxide-based solar disinfection is a very slow process that can have a decreased proportion of UV-A in sunlight⁴⁶. For the application of solar disinfection with titanium dioxide to improve visible light absorption of TiO₂ or UV-A, flourishing work in doping with variety of metals or nitrogen is essential. Bacterial death in the dark was also shown by the titanium dioxide nanomaterial, indicating the likelihood of some unknown mechanisms. Silver has been recognised for its antimicrobial influence since ancient times. In industrial applications, the application of silver nanoparticles (AgNPs) in medical services and external medicines is miscellaneous.

As an antimicrobial nanomaterial, silver nanoparticles have been used in advanced years. The material they use to decontaminated water is as follows: antimicrobial activity, health and development are efficient and wide-ranging. The Nano Silver absorbs wide range of ions from Ag into water binding to-SH groups and kills important enzymes. Silver nanoparticle (AgNPs) toxicity mediated on the rate of discharge of silver ions. Silver ion release is influenced by the shape, scale, padding and crystallographic facets⁴⁷⁻⁵⁰. The existing of omnipresent ligands strengthens their toxicity and their bioavailability. As a protection for pathogens in micro ceramic filters that can be widely used in developing countries, silver nanoparticles have been used. Zinc oxide nanoparticles (ZONPs) have been used in sunscreen lotions, variety of paints and enormous coatings due to the improved UV absorption potential and transparency of visible light. In a wide variety of bacteria, zinc oxide nanoparticles (ZONPs) show improved antibacterial property. Nevertheless, since, for example, the researchers obtained opposite results, the antibacterial role of nanoparticles of zinc oxide was not apparent. Explanations for the selective photocatalytic processing of H₂O₂ for antimicrobial action of zinc oxide (ZO) have been proposed. Although ZO and Zn⁺² nanoparticles show antibacterial property, water species may be very susceptible to dissolved zinc. In the treatment of portable water, ZONPs are limited because of the ease of their dissolution⁵¹⁻⁵⁵.

A quantity of nanomaterials is recognized as having a effective property of catalytic oxidation and commonly used as highly developed oxidation processes in the control of water pollution. Scientists in different countries are drawing comprehensive attention to TiO₂nanoparticles, Fe₃O₄nanoparticles, ZnONPs and many other nanomaterials. In various approaches to Fenton or Fenton-like oxidation, photocatalytic oxidation and sonocatalytic oxidation in water purification, these sophisticated and advanced nanomaterials have been synthesised. Therefore, breakthroughs in research into these emerging materials are predictable to bring greatest progress in water environmental protection engineering applications⁵⁶⁻⁵⁸.

TiO₂ Nanoparticles

Titanium dioxide (TiO₂) nanoparticles are an outstanding photocatalyst for aqueous solution dilapidation of refractory organics under UV light irradiation. Electrons jump onto the transmission band to form conduction band e- and leave holes (h⁺) when TiO₂ is irradiated by UV radiation. At least 3.0 eV is the electrical probable of the holes, which is much higher than that of typical oxidants. As a consequence, the hydroxyl group (OH⁻) or H₂O absorbed on the wide range of surface nanoparticles of TiO₂ may be oxidised with high activity produced can also react with O₂ and produce different free radicals of superoxide, such as O₂⁻ and HO₂^{59,60}. In the oxidation of most organic compounds and a small number of inorganic compounds into CO₂, H₂O and small inorganic molecules, these energetic free radicals plays an optimistic role⁶¹⁻⁶⁴.

ZnO Nanoparticles

"Nano-ZnO is also considered to be a possible photocatalyst as a semiconductor nanomaterial due to its promising photocatalytic activity, non-toxic and inexpensive character. Consequently, the depletion of organic compounds in waste water has been thoroughly studied by nano-ZnO. 'Oskoei et al. (2015) tested the degradation under UV irradiation of humic acid (HA) in aqueous solution by nano-ZnO. Nearly 98.95 percent of HA was extracted by 0.5 g/L nano-ZnO within 0.5 h under acidic conditions, according to experimental findings. Using the coprecipitation method for photocatalytic dye degradation'^{65,66}.

Fe₃O₄ Nanoparticles and Iron-Based Nanomaterials

Fe₃O₄ is recognised as a typical magnetic material with an inverse spinel structure, and owing to its unusual wonderful paramagnetic and catalytic nature, nanometer sized Fe₃O₄ has attracted extensive attention by multiple researchers. Nano-Fe₃O₄ has been commonly in heterogeneous Fenton oxidation (Fo) of a variety of organic water contaminants in recent years, with an advanced catalytic activity compared to iron-based materials of non-nanometer scale. In nano-Fe₃O₄, there are two Fe-valences (Fe⁺² and Fe⁺³). Fe⁺² and Fe⁺³ can also plays a significant responsibility in the reaction to Fenton. Hydroxyl radicals can be fashioned and unconfined via the Haber-Weiss mechanism in the Fenton process⁶⁷.

Metallic Nanomaterials

"Compared to coarse crystal materials, metallic nanomaterials have singular and special structures, including Fe, Co, Ni, Cu, Zn, Ag, etc. Nanoscale metallic materials have recently been synthesised and used in several fields"⁶⁸. However, only ZVI nanoparticles are primarily used among these metallic nanomaterials to get rid of contamination in wastewater through reduction either in laboratory experiment or in grassland remediation. 'Since nZVI are primarily and successfully functional to groundwater remediation in situ, nZVI has generally become a very beautiful research subject as a reduction agent for the removal of groundwater pollution'⁶⁹⁻⁷¹. The efficacy of nZVI in the re-

removal of different pollutants has been successfully established on a laboratory scale, and some investigations have

also shown that nZVI has been used for the direct and field-scale implementation of a polluted site⁷²⁻⁷⁵.

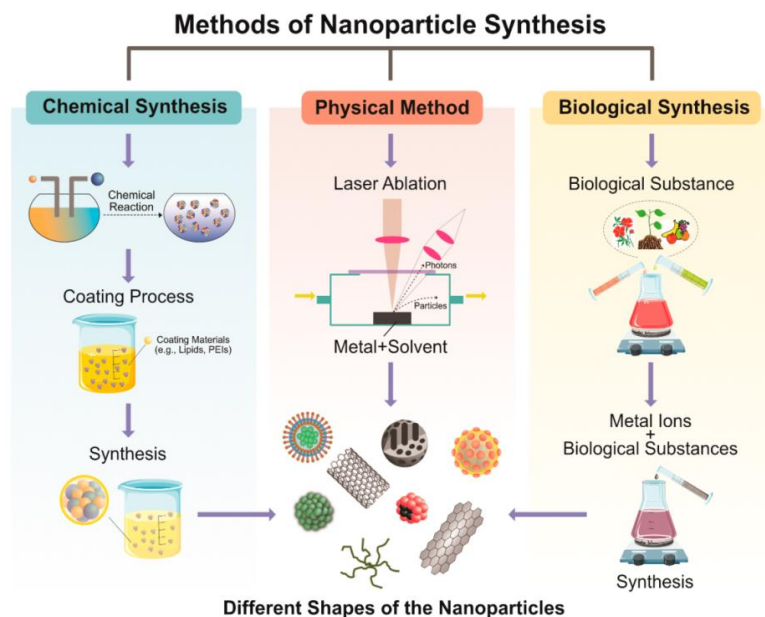


Figure 1: Methods of Nanoparticles Synthesis

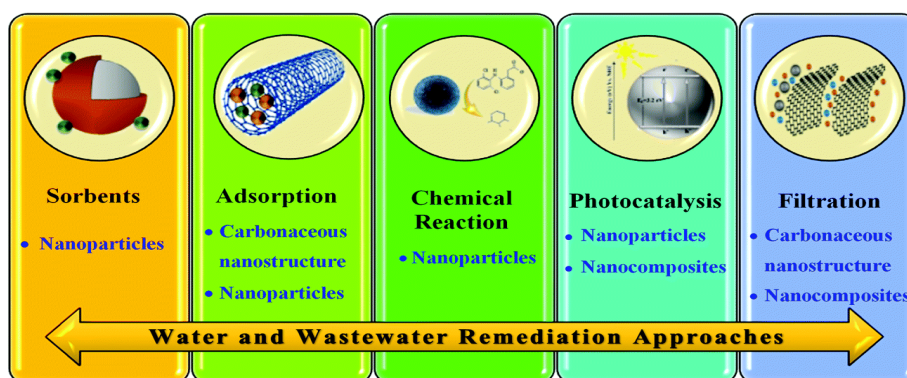


Figure 2: Different strategies of Wastewater Treatment

Outlook and Future Directions

For the subsequently generation of water sanitization strategies and systems to regulate water contamination, nanomaterials are promising building blocks. There are still some goals that need to be accomplished to speed up the realisation of nano-based water treatment approaches:

- (1) Next generation of nano adsorbents: In the production of nanoadsorbents with different surface functions, nanomaterials can be used to optimally absorb polar and non-polar water contaminants. Adsorption is combined with photodecomposition by using nano photocatalysts. Therefore, the organic pollutants are decaying into undisruptive by-products and, for a subsequent adsorption/photodecomposition period, the previously occupied surface is clean⁷⁶.
- (2) The next generation of nanomembranes: it is also possible to use nanomaterials as the construction of nanostructure membranes. Due to its modifiable pore size, astonishing porosity and open porous,

they allow the process of water treatment to be energy-efficient and thus economical. The given such merits, the commercial realisation of nanofibrous membranes has not yet taken place. Their capacity for water management, however, has been primarily demonstrated hypothetically rather than experimentally, and it will take time to practice such membranes⁷⁷⁻⁷⁹.

- (3) Energy effectiveness and scalability: many obstacles currently exist to the extensive industrial use of nanomaterials for the handling of water. These comprise technological limitations related to the scale-up and incorporation of different nanomaterials into the technology for water decontamination, protection, and cost efficiency. The most widely studied nanomaterials for dye adsorption are TiO₂ nanoparticles and carbon nanotubes (CNTs). However, their toxicity and expensive method of processing involving elevated tempera-

ture and pressure are also deterrent to industrialization^{80,81}.

- (4) Safe and eco-friendly nanomaterials: nanomaterials use in the production of micro filter, ultra-filter, and nanofilter membranes which be released into the water when the covering is exposed to violent streams of water with complex stress patterns. As a significant precedence, nanoparticles must therefore be firmly stabilised by physicochemical treatments on/in the membrane structure. In addition, nontoxic materials that are less environmentally challenging should be taken into account. The innovative invention of nanomaterials consequential from nature, such as cellulose based nanomaterials, may be shows potential in this regard^{82,83}.

Advancements in adsorption using nanomaterials

Procedures of Adsorbate using carbon-based nanomaterials are completely useful in the removal from water of wide organic and major inorganic matters. Adsorption is characterized as a outside occurrence in which material binds to a surface by conductivity induced primarily by electrostatic and van der Waals interactions from physical-chemical forces⁸⁴. A successful absorber must have a mixture of types characteristics, such as inert, excellent biocompatibility and mechanical strength resistant, and has to have a high ability to adsorb waste. These characteristics play an important role since they can evaluate the material's usefulness.

Adsorption procedures are based on a variety of factors including: pH, temperature, and concentration of pollutants, particle size, contact time and biosorbent and adsorption of physico-chemical nature. The pH can affect adsorption capabilities by changing the exterior groups here on the adsorbent and contaminant charge and improving the adsorption ability in endothermic reactions by raising temperatures. Adsorption of ibuprofen to activated carbon is supplementary beneficial at pH 3 than at pH 7. Furthermore, as the temperature rises to pH 3, ibuprofen adsorption has increased contact time (CT) with the pollutant may raise the quantity adsorbed, based on the adsorbent's surface and solution chemistry, as the length of time needed to saturate the adsorbent may differ. In assured system and not in additional systems a substance can therefore be a strong adsorbent⁸⁵⁻⁹⁰.

Sophisticated application of adsorbents

As described above, existing advanced filtration methods. Membrane processes are one such specialized method which, due to efficiency and reliability, have attracted immense popularity in EPC elimination. Membrane can be used to eliminate toxins from the water, as a division of microfiltration, ultra - filtration, nanofiltration, or reverse osmosis (RO) systems, by serving as a physical obstruction to pollutants^{91,92}.

Although extremely effective nanofiltration and inverted osmosis membranes, they demand greater energy than microfiltration and ultrafiltration membranes. Specif-

ic membrane technologies, such as ceramic membrane, polymers⁹³, metal-organic frames⁹⁴ and other specialized membranes are continually being investigated⁹⁵. Although a number of membranes survive, polyamide-based membranes (PBM) are among the majority frequently employed membranes for EPC removal⁹⁶⁻⁹⁸. Graphene and GO, for example, have been integrated into various polymer membranes to change their properties to build the process more efficient in waste management⁹⁹⁻¹⁰². The antimicrobial inhibition and enhancement of water flow through the membrane has been found in particular in GO-modified poly(N-vinylcarbazole) (PVC), polyamide and polysulfone membranes¹⁰³⁻¹⁰⁵.

2 CONCLUSION

For wastewater treatment, most of the nanotechnology-based technologies are widely innovating and promising strategies. The use of nanotechnology in a broad wastewater management programme is, however, not easy to introduce immediately. Wide range of applications alluded to here are either in vitro or set up in the laboratory. Because the application is for water control, extra care must be taken during its implementation to ensure protection and performance. As previously described, most nanomaterials are toxic substances. Inventive work in nanotechnology is an exciting new advancement for the treatment of wastewater, which is essential for individuals. These developments are quick, reliable and solid skills to treat wastewater by getting rid of unique types of water poisons. This paper revolves around the potential effects of nanotechnology in the treatment of wastewater. Nanoparticles, however, have another fundamental component that could make them an acceptable all-around technique: it is their capacity to identify and abstain from spoiling.

Acknowledgements

The authors would like to thank the Principal and Management of Sacred Heart College (Autonomous), Tirupattur, Tamil Nadu, India for rendering timely support and providing academic liberty for the successful completion of the review work.

Conflict of interest

The creators reported no irreconcilable situation. This report doesn't contain any investigations with human or creature subjects experienced by any of the authors.

References

- [1] M. I. Niyas Ahamed, S. Sathya, Ragul. V. An *in vitro* study on Hexavalent Chromium [Cr(VI)] Remediation using Iron Oxide Nanoparticles Based Beads. *Environmental Nanotechnology, Monitoring & Management*. 2020. 14: 1-5.
- [2] L. Mohammed, H. Gomma, D. Ragab, and J. Zhu. Magnetic nanoparticles for environmental and biomedical applications: Review. *Particuology*. 2016. 30: 1-8.
- [3] M. Barathi, A. S. K. Kumar, and N. Rajesh. Impact of fluoride in potable water - an outlook on the exist-

- ing defluoridation strategies and the road ahead. Coordination Chemistry Reviews. 2019. 387: 121-128.
- [4] M.I. Niyas Ahamed, V. Ragul, S. Anand, K. Kaviyarasu, V. Chandru and B. Prabhavathi. Green synthesis and toxicity assessment of nanozerovalent iron against chromium contaminated surface water. International Journal of Nanoparticles. 2018. 10(4): 312-325
- [5] Koteswar Rao M, Metre M. Effective low-cost adsorbents for removal of fluoride from water. International Journal of Science and Research. 2014. 3(6):120-124.
- [6] Yadav RK, Sharma S, Bansal M, Singh A, Pandey V, Maheshwari R. Effects of fluoride accumulation on growth of vegetable and crops in Dausa District Rajasthan. India. Adv Biores. 2016. 3 (4):14-16.
- [7] J R. Nagarajah, K. T. Wong, G. Lee. Synthesis of a unique nanostructured magnesium oxide coated magnetite cluster composite and its application for the removal of selected heavy metals. Separation and Purification Technology, 2017. 174: 290-300.
- [8] Zare, E.N., Motahari, A., Sillanpaa, M. Nano adsorbents based on conducting polymer nanocomposites with main focus on polyaniline and its derivatives for removal of heavy metal ions/dyes: a review. Environmental Research. 2018. 162: 173-195.
- [9] Peng X, Luan Z, Ding J, Di Z, Li Y, Tian B. Ceria nanoparticles supported nanotubes for the removal of arsenate from water. Mater Lett. 2005. 59:399-403.
- [10] Ashbolt, N. J. Microbial contamination of drinking water and disease outcomes in developing regions, J. Toxicology. 2004. 1(3): 229-238.
- [11] E. Bazrafshan, D. Balarak, A. Panahi, H. Kamani, and A. Mahvi. Fluoride removal from aqueous solutions by cupric oxide nanoparticle. Fluoride. 2016. 49: 233-244.
- [12] M. Zazouli, A. Mahvi, Y. Mahdavi, and D. Balarak. Isothermic and kinetic modeling of fluoride removal from water by means of the natural biosorbents sorghum and canola. Fluoride. 2015. 48: 1: 37-44.
- [13] Awaleh MO, Soubaneh YD (2014) Waste water treatment in chemical industries: the concept and current technologies. Hydrol Current Res 5:164
- [14] H. Akbari, F. Jorfi, A. Mahvi, M. Yousefi, and D. Balarak. Adsorption of fluoride on chitosan in aqueous solutions: determination of adsorption kinetics. Fluoride. 2018. 51(4): 319-327.
- [15] Nolan K. Copper toxicity syndrome. J Orthomol Med. 2003. 12:270-282
- [16] Abdelhafez, A.A., Li, J. Removal of Pb(II) from aqueous solution by using biochars derived from sugar cane bagasse and orange peel. J. Taiwan. Inst. Chem. Eng. 2016. 61: 367-375.
- [17] Baby R, Saifullah B, Hussein MZ. Carbon nanomaterials for the treatment of heavy metal-contaminated water and environmental remediation. Nanoscale Res Lett. 2019. 14:341
- [18] Abdolali, A., Ngo, H.H., Guo, W., Lu, S., Chen, S.S., Nguyen, N.C., Zhang, X., Wang, J., Wu, Y. A breakthrough biosorbent in removing heavy metals: equilibrium, kinetic, thermodynamic and mechanism analyses in a lab-scale study. Sci. Total Environ. 2016. 542: 603-611.
- [19] Ahmad, A., Ghazi, Z.A., Saeed, M., Ilyas, M., Ahmad, R., Khattaka, A.M., Iqbal, A. A comparative study of the removal of Cr(VI) from synthetic solution using natural biosorbents. New Journal of Chemistry. 2017. 41: 10799-10807.
- [20] Arora G, Bhateja S. Estimating the fluoride concentration in soil and crops grown over it in and around Mathura, Uttar Pradesh, India. Am J Ethno Med. 2014. 1(1):36-41
- [21] Fukushima M, Ishizaki A, Sakamoto M. On distribution of heavy metals in rice field soil in the "Itai-itai" disease epidemic district. Nihon Eiseigaku Zasshi. 1970. 24:526-535
- [22] Babel, S., and Kurniawan, T. A. Low-cost adsorbents for heavy metals uptake from contaminated water: A review. J. Hazardous Materials, 2003. 97: 219-243.
- [23] J. He, Y. An, and F. Zhang. Geochemical characteristics and fluoride distribution in the groundwater of the Zhangye Basin in Northwestern China. Journal of Geochemical Exploration. 2017. 135: 22-30.
- [24] Ashbolt, N. J. Microbial contamination of drinking water and disease outcomes in developing regions, J. Toxicology., 2004. 1(3): 229-238.
- [25] L. Bo, Q. Li, Y. Wang, L. Gao, X. Hu, and J. Yang. Adsorptive removal of fluoride using hierarchical flower-like calcined Mg-Al layered double hydroxides. Environmental Progress and Sustainable Energy. 2019. 35(5): 1420-1429.
- [26] Bai, C. Hu, H. Liu, and J. Qu. Selective adsorption of fluoride from drinking water using NiAl-layered metal oxide film electrode. Journal of Colloid and Interface Science. 2016. 539: 146-151.
- [27] Dong, Z., Zhang, F., Wang, D., Liu, X., Jin, J. Polydopamine-mediated surface-functionalization of graphene oxide for heavy metal ions removal. J. Solid State Chem. 2015. 224: 88-93.
- [28] Deliyanni, E. A., Bakoyannakis, D. N., Zouboulis, A. I. and Matis, K. A. Sorption of As(V) ions by akaganeite-type nanocrystals. Chemosphere, 2003. 50(1): 155-163.
- [29] Sheet I, Kabbani A, Holail H. Removal of heavy metals using nanostructured graphite oxide, silica nanoparticles and silica/ graphite oxide composite. Energy Proc 2014. 50:130-138
- [30] Kim, S., Chu, K., AlHamadani, Y., Park, C., Jang, M., Kim, D., Yu, M., Heo, J., Yoon, Y., Removal of contaminants of emerging concern by membranes in wa-

- terandwastewater:areview.Chem.Eng.J.2018. 335:896-914.
- [31] Krauklis, A., Ozola, R., Burlakovs, J., Rugele, K., Kirillov, K., Trubaca-Boginska, A., Rubenis, K., Stepanova, V., Klavins, M. FeOOH and Mn₈O₁₀Cl₃ modified zeolites for As(V) removal in aqueous medium. *J. Chem. Technol. Biotechnol.* 2017. 92, :1948-1960.
- [32] Deliyanni, E. A., Bakoyannakis, D. N., Zouboulis, A. I. and Matis, K. A. Sorption of As(V) ions by akaganeite-type nanocrystals. *Chemosphere*, 2003. 50(1): 155-163.
- [33] Ding, Z., Hu, X., Morales, V.L., Gao, B. Filtration and transport of heavy metals in graphene oxide enabled sand columns. *Chem. Eng. Journal*, 2014. 257: 248–252.
- [34] Obare SO, Meyer GJ. Nanostructured materials for environmental remediation of organic contaminants in water. *Journal of Environmental Science and Health A*, 2004. 39:2549–2582
- [35] Kumarasinghe, U., Inoue, Y., Saito, T., Nagamori, M., Sakamoto, Y., Mowjood, M., Kawamoto, K. Temporal variations in perched water and groundwater qualities at an open solid waste dumpsite in Sri Lanka. *Int. J. of. Geomate.* 2017.13(38):01-08.
- [36] Boronina, T., Klabunde, K. and Sergeev, G. Destruction of organohalides in water using metal particles-carbon tetrachloride/water reactions with magnesium, tin and zinc. *Env. Sci. Technol.* 1995. 29(6): 1511-1517.
- [37] Nawab, J., Khan, S., Ali, S., Sher, H., Rahman, Z., Khan, K., Tang, J., Ahmad, A. Health risk assessment of heavy metals and bacterial contamination in drinking water sources: a case study of Malakand Agency. Pakistan. *Environ. Monit. Assess.* 2016. 188(5): 286-297.
- [38] Bhattacharya, S., Saha, I., Mukhopadhyay, A., Chattopadhyay, D., Ghosh, U. C. and Chatterjee, D. Role of nanotechnology in water treatment and purification: Potential applications and implications. *International Journal of Chemical Science Technology.* 2003. 3(3): 59-64.
- [39] Nwab, J., Khan, S., Khan, M., Sher, H., Rehman, U., Ali, S., Shah, S. Potentially toxic metals and biological contamination in drinking water sources in chromite mining-impacted areas of Pakistan: a comparative study. *Expo. Health.* 2017. 9, 275-287.
- [40] Savage, N., and Diallo, M. S. Nanomaterials and water purification: Opportunities and challenges. *Journal of Nanoparticle Research.* 2005. 7: 331-342
- [41] H. A. Dharmagunawardhane, S. P. K. Malaviarachchi, and W. Burgess. Fluoride content of minerals in gneissic rocks at an area of endemic dental fluorosis in Sri Lanka: estimates from combined petrographic and electron microprobe analysis. *Ceylon Journal of Science.* 2016. 45(1): 57–66.
- [42] Nris, J., Luzardo, F., Silva, E., Velasco, F. Evaluation of adsorption processes of metal ions in multi-element aqueous systems by lignocellulosic adsorbents applying different isotherms: a critical review. *Chem. Eng. J.* 2019. 357:404-420.
- [43] Park, S. and Ruoff, R. S. Chemical methods for the production of graphenes. *Nature Nanotechnol.* 2009. 4: 217-224
- [44] Manna, B., Dasgupta, M and Ghosh, U. C. Crystalline hydrous titanium (IV) oxide (CHTO): An arsenic (III) scavenger. *J. Water Supply Res. Technol.* 2004. 53(7): 483-495.
- [45] Zhang, W. X. Nano-scale iron particles for environmental remediation: an overview. *Journal of Nanoparticle research.* 2003. 5: 323-332
- [46] Leonard, P., Hearty, S and Brennan, J. Advances in biosensors for detection of pathogens in food and water. *Enzyme and Microbial Technol.* 2003. 32(1): 3-13..
- [47] Shah, M. A., and Ahmed, T. Principles of Nanoscience and Nanotechnology. Narosa Publishing House, New Delhi, India, 2011. 34- 47.
- [48] Amin MT, Alazba AA, Manzoor U. A review of removal of pollutants from water/wastewater using different types of nanomaterials. *Journal of Advance Material Science and Engineering.* 2014:1–24
- [49] Hu, W., Peng, C., Luo, W., Lv, M., Li, X., Li, D., Huang, Q and Fan, C. Graphene-based antibacterial paper. *ACS Nanotechnology.* 2010. 4: 4317- 4323.
- [50] Mamadou, S. D., and Savage, N. Nanoparticles and water quality. *Journal of Nanotechnology Research*, 2005. 7: 325-330.
- [51] Sreeprasad, T. S., Maliyekkal, S. M., Lisha, K. P. and Pradeep, T. Reduced graphene oxide-metal/metal oxide composites: Facile synthesis and application in water purification. *J. Hazardous Materials*, 2011.186: 921-931.
- [52] Akasaka T, Watari F. Capture of bacteria by flexible carbon nanotubes. *Acta Biomater* 2009. 5:607–612
- [53] Obare, S. O., and Meyer, G. J. Nanostructured materials for environmental remediation of organic contaminants in water. *Journal of Environmental Science Health.* 2004. 39(10): 2549- 2582.
- [54] Sulyman, M., Namiesnik, J., Gierak, A. Low-cost adsorbents derived from agricultural by-products/wastes for enhancing contaminant uptakes from wastewater: a review. *Pol. Journal of Environmental Study* 2017. 26 (2): 479-510.
- [55] G. Mishra, B. Dash, and S. Pandey. Layered double hydroxides: a brief review from fundamentals to application as evolving biomaterials. *Applied Clay Science.* 2018. 153: 172–186.
- [56] Tiwari, D. K., Behari, J. and Sen, P. Applications of nanoparticles in wastewater treatment. *Journal of World Applied Science.* 2008. 3(3): 417- 433.
- [57] U. Maheshwari, Removal of metal ions from wastewater using adsorption: experimental and theoretical studies, Ph.D. thesis, Birla Institute of Technology and Science, Pilani, Rajasthan, India, 2015.

- [58] Zhang, K., Dwivedi, V., Chi, C. and Wu, J. Graphene oxide/ferric hydroxide composites for efficient arsenate removal from drinking water. *Journal of Hazardous Materials*, 2010. 182: 162- 168.
- [59] Abdolali, A., Ngo, H.H., Guo, W., Lu, S., Chen, S.S., Nguyen, N.C., Zhang, X., Wang, J., Wu, Y. A breakthrough biosorbent in removing heavy metals: equilibrium, kinetic, thermodynamic and mechanism analyses in a lab-scale study. *Science Total Environment*, 2016. 542: 603-611.
- [60] Ahmad, A., Ghazi, Z.A., Saeed, M., Ilyas, M., Ahmad, R., Khattaka, A.M., Iqbal, A. A comparative study of the removal of Cr(VI) from synthetic solution using natural biosorbents. *New J. Chem.* 2017. 41: 10799-10807.
- [61] Arora G, Bhateja S. Estimating the fluoride concentration in soil and crops grown over it in and around Mathura, Uttar Pradesh. India. *American Journal of Ethno Medicine*. 2014. 1(1):36-41
- [62] Zhang, S., Li, X. Y. and Chen, J. P., 2010, Preparation and evaluation of a magnetite doped activated carbon fiber for enhanced arsenic removal. *J. Carbon.*, 48: 60-67.
- [63] J. He, Y. An, and F. Zhang. Geochemical characteristics and fluoride distribution in the groundwater of the Zhangye Basin in Northwestern China. *Journal of Geochemical Exploration*. 2013. 135: 22-30.
- [64] L. Mohammed, H. Gomma, D. Ragab, and J. Zhu. Magnetic nanoparticles for environmental and biomedical applications: Review. *Particuology*. 2016. 30: 1-8.
- [65] M. Barathi, A. S. K. Kumar, and N. Rajesh. Impact of fluoride in potable water - an outlook on the existing defluoridation strategies and the road ahead. *Coordination Chemistry Reviews*. 2019. 387: 121-128.
- [66] Chen, M.; Zhu, L.; Dong, Y.; Li, L.; Liu, J. Waste-to-Resource Strategy To Fabricate Highly Porous Whisker-Structured Mullite Ceramic Membrane for Simulated Oil-in-Water Emulsion Wastewater Treatment. *ACS Sustainable Chemistry & Engineering* 2016. 4: 2098-2106.
- [67] Adeleye AS, Engineered nanomaterials for water treatment and remediation: costs, benefits, and applicability. *Chemistry Engineering Journal*, 2010. 286:640-662
- [68] Bhaumik M. Composite nanofibers prepared from metallic iron nanoparticles and polyaniline: high performance for water treatment applications. *Journal of Colloid Interface Science*, 2014. 425: 75-82
- [69] Carpenter AW, de Lannoy CF, Wiesner MR. Cellulose nanomaterials in water treatment technologies. *Environ Sci Technology*. 2015. 49(9):5277-5287
- [70] Deng N. Discoloration of aqueous reactive dye solutions in the UV/Fe-0 system. *Water Research*. 2000. 34(8):2408-2411
- [71] Deng L. SnS₂/TiO₂ nanocomposites with enhanced visible light-driven photoreduction of aqueous Cr(VI). *Ceram Interface*. 2016. 42(3):3808-3815
- [72] Kritis AA. Latest aspects of aldosterone actions on the heart muscle. *J Physiol Pharmacology*. 2016. 67(1):21-30
- [73] Kahrizi H, Bafkar A, Farasati M. Effect of nanotechnology on heavy metal removal from aqueous solution. *Journal of Center South University*, 2016. 23:2526-2535
- [74] Kurian M, Nair DS. Heterogeneous Fenton behavior of nano nickel zinc ferrite catalysts in the degradation of 4-chlorophenol from water under neutral conditions. *J Water Process Engineering*, 2014. 8:37-49
- [75] Ma J. Fabrication of Ag/TiO₂ nanotube array with enhanced photo-catalytic degradation of aqueous organic pollutant. *Physica E Low-Dimensional Systematic Nanostructure*. 2014. 58:24-29
- [76] Li YH, Wang S, Wei J, Zhang X, Xu C, Luan Z, Wu D. Lead adsorption on carbon nanotubes. *Chemistry Physical Letter*. 2002. 3:263-266
- [77] Ma BW. Modification of ultrafiltration membrane with nanoscale zerovalent iron layers for humic acid fouling reduction. *Water Research*. 2015. 71:140-149
- [78] Oskoei V. Removal of humic acid from aqueous solution using UV/ZnO nano-photocatalysis and adsorption. *Journal of Molecular Liquid*. 2015. 213:374-380
- [79] Lu C, Liu. Removal of nickel (II) from aqueous solution by purified carbon nanotubes. *Journal of Chemical Technology Biotechnology*, 2006. 81:1932-1940
- [80] Ravikumar KVG. A comparative study with biologically and chemically synthesized nZVI: applications in Cr(VI) removal and ecotoxicity assessment using indigenous microorganisms from chromium-contaminated site. *Environmental Science Pollution Research*, 2016. 23(3):2613-2627
- [81] Akasaka T, Watari F. Capture of bacteria by flexible carbon nanotubes. *Acta Biomaterial*. 2009. 5:607-612
- [82] Ruzhitskaya O, Gogina E. Methods for removing of phosphates from wastewater. *MATEC Web Conference*. 2017. 106:67
- [83] Kumar R, Chawla J. Removal of cadmium ion from water/ wastewater by nano-metal oxides. *Water quality exposure and health*, 5th edn. Springer, New York. 2014.
- [84] Salem HM, Eweida EA, Farag A. Heavy metals in drinking water and their environmental impact on human health. In: *ICEHM2000*. 2000. 542-556
- [85] Ambrose AM, Larson DS, Borzelleca JR, Hennigar GR Jr. Long-term toxicological assessment of nickel in rats and dogs. *Journal of Food Science Technology*. 1976. 13:181-187
- [86] Ambrose AM, Larson DS, Borzelleca JR, Hennigar GR Jr. Long-term toxicological assessment of nickel

- el in rats and dogs. *Journal of Food Science Technology*. 1976. 13:181–187
- [87] Amin MT, Alazba AA, Manzoor U. A review of removal of pollutants from water/wastewater using different types of nanomaterials. *Journal of Advance Material Science Engineering*. 2014:1–24
- [88] Huang J, Cao Y, Liu Z, Deng Z, Wang W. Application of titanatenanofowers for dye removal: a comparative study with titanate nanotubes and nanowires. *Chemical Engineering Journal*. 2012. 191:38–44
- [89] Iijima S. Helical microtubules of graphitic carbon. *Nature*. 1991. 354:56–58
- [90] Iijima S, Ichihashi T. Single-shell carbon nanotubes of 1-nm diameter. *Nature* 1993. 363:603–605
- [91] Inoue Y, Hoshino M, Takahashi H, Noguchi T, Murata T, Kanzaki Y, Hamashima H, Sasatsu M. Bactericidal activity of Agzeolite mediated by reactive oxygen species under aerated conditions. *Journal of Inorganic Biochemistry*. 2002. 92:37–42
- [92] Kocabas ZO, Aciksoz B, Yurum Y. Binding mechanisms of As(III) on activated carbon/titanium dioxide nanocomposites: a potential method for arsenic removal from water MRS online proceedings library, 2012. 1449. Cambridge University Press, Cambridge
- [93] Jain P, Pradeep T (2005) Potential of silver nanoparticle-coated polyurethane foam as an antibacterial water filter. *Biotechnology Bioengineering*. 2005. 90:59–63
- [94] Khan K, Rehman S (2016) Wonders of Nanotechnology. In: Khan SB, Asiri AM, Akhtar K (eds) Development and prospective applications of nanoscience and nanotechnology (Nanomaterials and their Fascinating Attributes), 1st edn. Bentham Publishers, UK Khaydarov RA, Khaydarov RR, Gapurova O. Water purification from metal ions using carbon nanoparticle-conjugated polymer nanocomposites. *Water Research*. 2010. 44:1927–1933
- [95] Jeon HJ, Kim JS, Kim TG, Kim JH, Yu WR, Youk JH. Preparation of poly-caprolactone based polyurethane nanofibers containing silver nanoparticles. *Applied Surface Science*. 2008. 254:5886–5890
- [96] Yang Y, Zhang H, Wang P, Zheng Q, Li J. The influence of nanosized TiO₂ fillers on the morphologies and properties of PSF UF membrane. *Journal of Membrane Science*. 2007. 288:231–238
- [97] Jaishankar M, Tseten T, Anbalagan N, Mathew BB, Beeregowda KN. Toxicity, mechanism and health effects of some heavy metals. *Interdiscip Toxicology*. 2014. 7(2):60–72
- [98] Zhao G, Stevens SE Jr. Multiple parameters for the comprehensive evaluation of the susceptibility of *Escherichia coli* to the silver ion. *Biometals*. 1998. 11:27–32
- [99] Kaaber K, Vein N, Tjell JC. Low nickel diet in the treatment of patients with chronic nickel dermatitis. *Br Journal of Dermatology*. 1978. 98:197–201
- [100] Liu X, Wang M, Zhang S, Pan B. Application potential of carbon nanotubes in water treatment: a review. *Journal of Environmental Science*. 2013. 25:1263–1280
- [101] Kumar M, Puri A. A review of permissible limits of drinking water. *Indian J Occup Environ Med*. 2012. 16:40–44
- [102] Kumar S, Nehra M, Mehta J, Dilbaghi N, Marrazza G, Kaushik A. Point-of-care strategies for detection of waterborne pathogens. *Sensors*. 2019. 19(20):4476
- [103] Lala NL, Ramaseshan R, Li B, Sundarrajan S, Barhate RS, Liu YJ, Ramakrishna S. Fabrication of nanofibers with antimicrobial functionality used as filters: protection against bacterial contaminants. *Biotechnol Bioengineering*. 2007. 97:1357–1365
- [104] Amin, M. T., Alazba, A. A. and Manzoor, U., 2014. A review of removal of pollutants from water/wastewater using different types of nanomaterials. *Advance Material Science and Engineering*. 2014. 1(5):125–146
- [105] Liu J, Ma Y, Xu T, Shao G. Preparation of zwitterionic hybrid polymer and its application for the removal of heavy metal ions from water. *Journal of Hazard Material*. 2010. 178:1021–1029